

Fluoride Release and Uptake into Hydroxyapatite from Experimental Dental Varnish

I. Ahmed*, S.S. Coleman, C.M. Carey

1:University of Colorado, Anschutz Medical Campus, School of Dental Medicine, Aurora, CO, USA



Abstract

Fluoride releasing dental varnishes (F-varnish) typically contain 5% NaF within a resin or rosin base. The sodium fluoride concentration ([NaF]) that yields optimal fluoride enamel uptake, release, and anti-cares efficacy is unknown. Also, there is controversy on whether the release rate of fluoride uptake into enamel. Objectives: measure fluoride uptake (loosely-and tightly-bound) into hydroxyapatite (HAp) discs and fluoride release as a function of [NaF] in F-varnish. Methods: triplicate sets of F-varnish containing 5.0%, 2.5%, 1.25%, and 0.625% (wt/wt) NaF-were prepared. 0.01-0.02g F-varnish was applied to one side of a HAp disc. Fluoride release into a continuous flow of 30mmol/L KCI at 1 mL/min was measured over 3h. The HAp discs were the cleaned and loosely-bound fluoride extracted overnight in KOH (1.0mL, 1.0mol/L). Tightly bound fluoride was extracted via serial extractions in HClO₄ (1.0mL, 0.5mol/L). The extraction solutions were neutralized and analyzed via fluoride ion-selective electrode. Results: the cumulative fluoride release profile increased hyperbolicality over 3h. The total fluoride released at 3h ([F]₁₈₉) increased as a function of [NaF]. The time for 50% (t_{sc}) of the fluoride to be released as \$32min for all varnishes. Loosely- and tightly-bound fluoride increased to 2.5% NaF and plateaued.

[NaF]%	[F] Tightly Bound (µg/mL)	[F] Loosely Bound (µg/mL)	[F] ₁₈₀ released (µgF/mg Varnish)	t ₅₀ (minutes)
0.625	0 10(0 03)*	0 09(0 05)	0.7 (0 23)	4.40
1.25	0.15(0.02)	0.26(0.12)	1.87(0.86)0	31.95
2.5	0.25(0.03)°	1.20(0 03)*	2.47(0 47)3	26.10
5.0	0.19(0.05)	1.18(0.07)*	3.82(0.84)5	19.70

Values in the table are mean(standard error).

Superscripts indicate groups that are not significantly different in each category (ANOVA, Newman-Keuls, p<0.05).

Conclusions: In this study, optimal [NaF] is 2.5% for saturating all the available tight and loose binding sites of the HAp disc. Fluoride release into solution is not a reliable indicator of fluoride uptake. Further studies using enamel are planned.

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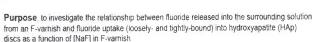
Introduction

What do we know about F-Varnish?

- a ADA recommends for caries prevention, at-risk patients On the basis of critical reviews of clinical literature
- US FDA does not recognize F-varnishes for caries prevention Research has not proven that F-varnishes can prevent caries Approved for cavity liners and treatment of hypersensitivity
- c. Numerous F-varnishes in the marketplace
- d. Marketed for more than 30 years
- e. Remains on teeth for 2 to 24 hours
- f. 50,000 ppm NaF (22,600 ppm F)
- g. 6 month efficacy

What don't we know?

- a Mechanism of action
- b. Optimal application frequency
- c. Optimal application conditions
- d. Optimal product composition for efficacy
- e. If release rate is predictive of efficacy
- f. Identity of the significant factors that affect efficacy
- g Short term and long term effects on patients



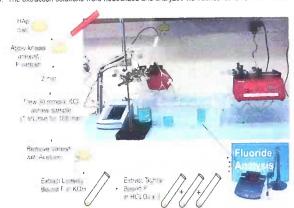


Hypotheses

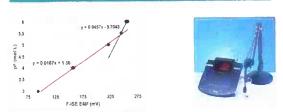
- There is a fluoride saturation point where the tooth can not take up anymore fluoride
 Meaning: Additional fluoride will not increase anti-caries efficacy
- 2. Fluoride release into saliva is not predictive of the Fluoride uptake into the tooth
- Meaning: Fluoride uptake by the enamel is INDEPENDENT of the fluoride release from the varnish

Experimental Design

- 1 F-free varnish was used to prepare 4 different [NaF] varnishes via serial dilutions: 5%, 2.5%, 1.25%, 0.625% (w/w)
- 2 Hydroxyapatite discs (mimic tooth composition), were coated on one side with the experimental varnish
- Fluoride release into a continuous flow of 30mmol/L KCI (mimics salivary ionic strength) at 1 ml /min was measured over 3h
- 4 The HAp discs were then cleaned with acetone.
 - 1 Loosely bound fluoride extracted overnight in KOH (1.0mL, 1.0mol/L)
- 2. Tightly bound fluoride was extracted via serial extractions in HClO4 (1 0mL, 0.5mol/L).
- 5. The extraction solutions were neutralized and analyzed via fluoride ion-selective electrode



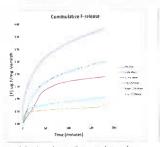
Analysis for Fluoride

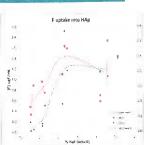


F-Extraction

Calculate the fluoride concentration of the sample with the appropriate standard curve. NOTE: F-ISE is not as sensitive at the low [F] thus requiring the use of the non-ideal curve for pF 5.3 to pF 6.0.

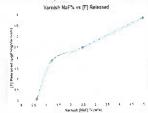
Results





The following observations can be made.

- . The cumulative amount of F released into the solution increased asymptotically over time
- The total amount of F release is proportional to the NaF concentration in the varnish on the HAN substrate.
- Both the loosely bound and tightly bound F uptake into the HAp disc as a function of varnish NaF concentration is non-linear
- There is an optimal NaF concentration at 2.5 % that saturates HAp fluoride binding sites.



- The time where 50% of the fluoride to be released was ≤ 32 min for all experimental varnishes
- The amount fluoride released into saliva is not a reliable indicator of fluoride uptake
- Preliminary data shows that the [F]₁₈₀ released from enamel discs was 1.66 (0.27) µg F/mg varnish, n=2 which is significantly different from that of the HAp discs at 2.5 % NaF (p < 0.05)

Discussion

- · Tightly bound Fluoride gives long term protection by preventing dissolution
- Loosely bound fluoride gives short term protection by inducing mineralization and through
- The concentration of fluoride binding sites on the HAp disc is limited.
- · Saturation of fluoride binding sites gives maximal protection against demineralization (caries)
- Research indicates a 6-month efficacy for caries protection, thus the therapy should be repeated twice annually for at-risk patients
- At %'s lower than 5 % NaF, a fluoride varnish can saturate the HAp disc

Conclusions

- In this study, optimal [NaF] is 2.5% for saturating all the available tight and loose binding sites of the HAp disc
- More experiments are needed to identify compositional factors which may be significant
- · Experiments using enamel instead of HAp discs are needed

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